

PROVISIONAL  
INTERNATIONAL UNION OF PURE AND APPLIED  
CHEMISTRY  
and  
INTERNATIONAL UNION OF BIOCHEMISTRY  
JOINT COMMISSION ON BIOCHEMICAL NOMENCLATURE\*

**NOMENCLATURE OF  
BRANCHED-CHAIN  
MONOSACCHARIDES**

Comments on these recommendations are welcome and should be sent within 8 months from January 1982 to the Secretary of the Commission

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Cambridge CB2 1QW  
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Comments from the viewpoint of languages other than English are especially encouraged. These may have special significance regarding the publication in various countries of translations of the nomenclature eventually approved by IUPAC.

\*Membership of the Commission for 1979-81 was as follows:

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*Expert Panel:* H. PAULSEN (Convener, FRG); L. C. CROSS (UK); late E. HARDEGGER (Switzerland); O. HOFFMANN-OSTENHOF (Austria); D. HORTON (USA); K. L. LOENING (USA); D. J. MANNERS (UK); W. G. OVEREND (UK); R. S. TIPSON (USA).

## IUPAC-IUB Joint Commission on Biochemical Nomenclature (JCBN)

## Nomenclature of Branched-Chain Monosaccharides

## Recommendations 1980

The naming of acyclic forms is described, and the first part of this is the rules for choosing which is the parent chain. This leads to naming the parent monosaccharide, including the naming of the C-substituent that forms the branch. Stereochemical conventions have to be extended to specify configuration: the two substituents at one carbon on the parent chain are put in order of priority; that of higher priority is considered to replace hydroxyl and that of lower priority to replace hydrogen in assigning a name to the parent monosaccharide. Equivalent groups and terminal substitution are dealt with separately. Cyclic forms are then described. Many examples are given.

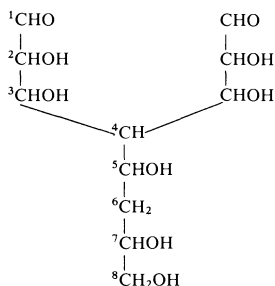
## 1. NAMING THE ACYCLIC FORM

## 1.1. Basis of the Name

A branched-chain monosaccharide will be named as a derivative of a parent unbranched monosaccharide, expressed in its acyclic form.

## 1.2. Choice of Parent Chain

The parent chain will be the all-carbon unbranched chain having one or more potential carbonyl groups and containing the greatest number of monosaccharide carbon units. If any choice remains, preference will be given to the chain containing the greatest number of carbon atoms. Example:

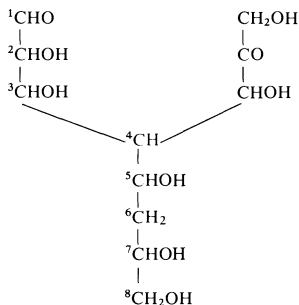


Document of the IUB-IUPAC Joint Commission on Biochemical Nomenclature (JCBN). Its members are: P. Karlson (chairman), H. B. F. Dixon, B. L. Horecker, Y. Jeannin, C. Liébecq (as chairman of the IUB Committee of Editors of Biochemical Journals), B. Lindberg, K. L. Loening, and G. P. Moss. Comments and suggestions for future revisions of these recommendations may be sent to its secretary, H. B. F. Dixon, University Department of Biochemistry, Tennis Court Road, Cambridge, England, CB2 1QW, or any member. JCBN thanks the expert panel of H. Paulsen (convener), L. C. Cross, E. Hardegger, O. Hoffmann-Ostenhof, D. Horton, K. L. Loening, D. J. Manners, W. G. Overend, and R. R. Tipson for drafting these proposals, its predecessor, CBN, for setting up the panel, and other members of the Nomenclature Committee of IUB (H. Bielka, W. B. Jakoby, B. Keil, and E. C. Webb) for consultation.

The parent is the numbered octose, not the alternative heptodialdose.

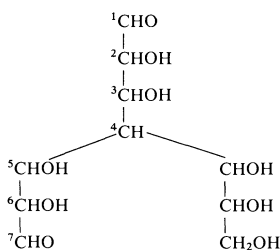
If choice remains after the application of the above rule, that choice will be made according to the following order of preference, treated in the order given.

a) An aldose has preference over a ketose. Example:



The parent is the numbered octose, not the alternative 2-octulose.

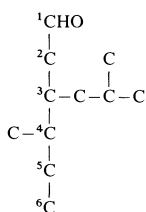
b) The maximum number of (potential) carbonyl groups. Example:



The numbered dialdose has preference over the alternatives.

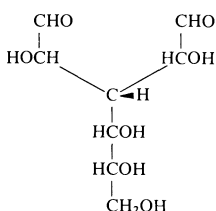
Reproduced from *Eur. J. Biochem.*, Vol.119, pp.5-8 (1981) by courtesy of Springer-Verlag, Heidelberg, FRG.

c) The branch point(s) will have the lowest locants possible. Example:

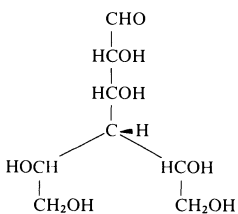


The parent is the vertical numbered chain, not the alternative turning right at 3, because the locants of the branches are 3,4, instead of 3,5.

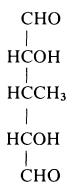
d) The priority of rule Carb-2 (here and elsewhere see [1]). Examples:



3-Deoxy-3-C-(L-glycero-1-hydroxy-2-oxoethyl)-D-altrio-hexose (not D-glycero-, D-gluco- by Rule Carb-2a)



4-Deoxy-4-C-(L-glycero-1,2-dihydroxyethyl)-D-gulo-hexose (not D-glycero-, L-talo- by Rule Carb-2a)

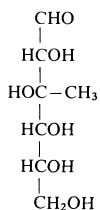


3-Deoxy-3-methyl-ribo-pentodialdose. N.B. Considered as the D-form for defining the direction of numbering (Rule Carb-2b)

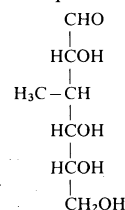
### 1.3. Naming of Parent Monosaccharide

Unless the two substituents at the branch point of the parent chain are equivalent, or one of them is equivalent to an attached portion of the parent chain (see 1.4), it is necessary to distinguish between these two substituents in order to specify the stereochemistry of the branch point. The priority of the two substituents (including hydrogen if present) at the branch point is determined by the procedure (based largely on atomic number) used in the sequence rules (see Appendix 2 in [3]). The configuration symbol is then derived from Rule Carb-8 using the substituent having the higher priority in place of the corresponding hydroxyl group

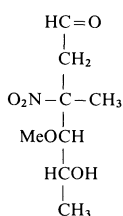
(cf. Rule Carb-14 for amino sugars and Rule Carb-36 for sulfur and selenium analogues). Examples:



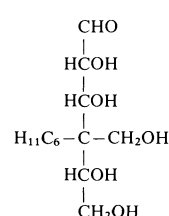
3-C-Methyl-D-glucose (oxygen has priority over carbon)



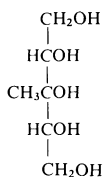
3-Deoxy-3-C-methyl-D-gluco-hexose (carbon has priority over hydrogen)



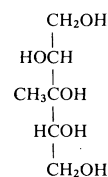
2,3,6-Trideoxy-3-C-methyl-4-O-methyl-3-C-nitro-D-lyxo-hexose (nitrogen has priority over carbon)



4-C-Cyclohexyl-4-deoxy-4-C-(hydroxymethyl)-D-allo-hexose (oxygen has priority over carbon [C-2'])



3-C-Methyl-ribitol

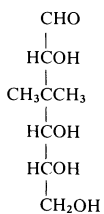


3-C-Methyl-D-arabinitol

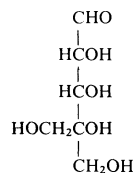
Note. The first example is named from D-glucose because it is a substituted D-glucose; the second is named from D-gluco-hexose because by loss of oxygen it has ceased to be a substituted glucose although it has the D-gluco configuration.

### 1.4. Naming of Parent Monosaccharides that Have Two Equivalent Groups

If the two substituents at the branch point of the parent chain are equivalent, or if one of them is equivalent to an attached portion of the parent chain, then the parent monosaccharide must be named according to Rule Carb-14 for unsubstituted deoxymonosaccharides. Examples:



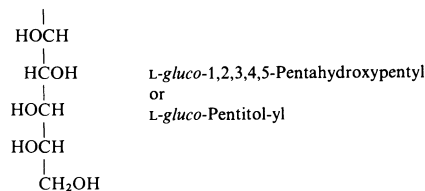
3-Deoxy-3,3-di-C-methyl-D-ribo-hexose



4-C-(Hydroxymethyl)-D-erythro-pentose

### 1.5. The Branches

Each branch will be named either systematically or trivially as an alkyl or substituted alkyl radical, etc., replacing a C-hydrogen atom of the parent unbranched-chain monosaccharide. Within the branch the configurations around asymmetric centres will be indicated by use of configurational prefixes (Rules Carb-8 and Carb-9). If the first atom of the branch is chiral, the main chain counts as group X in applying Rule Carb-8. Example:



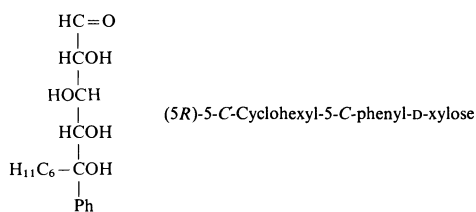
### 1.6. Numbering

The parent unbranched-chain monosaccharide will be numbered according to Rule Carb-4; i.e. in an aldose, the carbon atom of the (potential) carbonyl group is atom number one; in a ketose it has the lower number possible.

The carbon atoms of each branch will be numbered by Arabic numerals starting at the carbon atom attached to the main chain (see Nomenclature of Organic Chemistry, A-2-9).

## 2. TERMINAL SUBSTITUTION

If the terminal carbon atom of a monosaccharide is asymmetric as a result of the presence of two carbon substituents that cannot be regarded as belonging to the sugar chain, the stereochemistry at the new asymmetric centre will be indicated by use of the Sequence Rule (*R* or *S*) (see also Rules Carb-34 and Carb-35). Example:



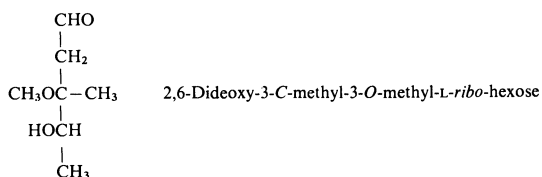
## 3. NAMING OF CYCLIC FORMS

In contrast with Rule Carb-20, the name of a cyclic form of a branched-chain monosaccharide will be based on that of the systematic name of the acyclic form wherein the two carbohydrate atoms between which cyclization has taken place are both considered to be part of the main chain. The cyclic form is then defined by the use of conventional carbohydrate configurational nomenclature.

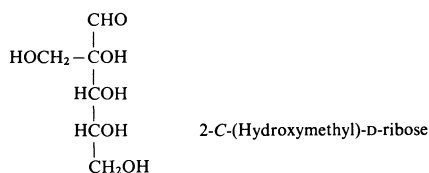
## 4. GENERAL EXAMPLES

### Acyclic Forms

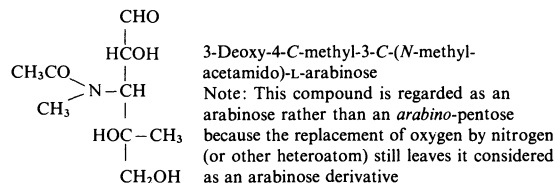
#### Cladinose



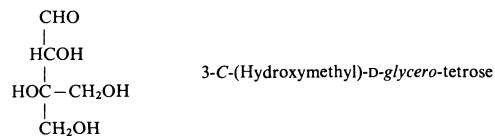
#### Hamamelose



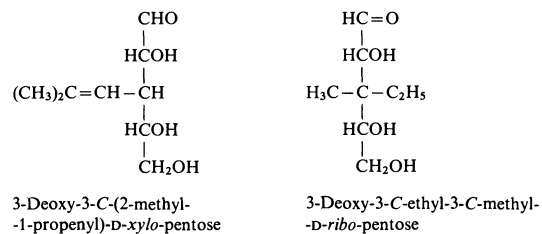
#### N-Acetylgarosamine



#### D-Apiose

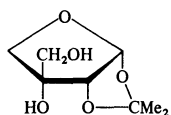
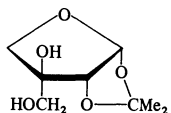


#### Other

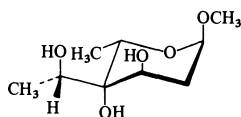


## Cyclic Forms

From D-apiose

3-C-(Hydroxymethyl)-1,2-O-isopropylidene- $\alpha$ -D-erythro-tetrahydrofuranose3-C-(Hydroxymethyl)-1,2-O-isopropylidene- $\beta$ -L-threo-tetrahydrofuranose

## Other

Methyl 2,6-dideoxy-4-C-(L-glycero-1-hydroxyethyl)- $\alpha$ -L-xylo-hexopyranoside

## REFERENCES

1. IUPAC Commission on the Nomenclature of Organic Chemistry and IUPAC-IUB Commission on Biochemical Nomenclature, Tentative rules for carbohydrate nomenclature, part 1, 1969, *Biochemistry*, 10, 3985–4004 and 4995 (1971), *Biochim. Biophys. Acta*, 244, 223–302 (1971), *Eur. J. Biochem.* 21, 455–476 (1971) and 25, 4 (1972) and *J. Biol. Chem.* 247, 613–635 (1972). See also pages 174–195 in [2].
2. International Union of Biochemistry (1978) *Biochemical Nomenclature and Related Documents*, The Biochemical Society, London.
3. IUPAC Commission on Nomenclature of Organic Chemistry, Rules for the nomenclature of organic chemistry, section E, stereochemistry (recommendations 1974) *Pure Appl. Chem.* 45, 11–30 (1976). See also pages 1–18 in [2] and section E in [4].
4. International Union of Pure and Applied Chemistry (1979) *Nomenclature of Organic Chemistry, Sections A, B, C, D, E, F and H*, Pergamon Press, Oxford.

### ERRATA

Volume, Issue and Year	Page no. and location	Correction				
49, 5 (1977)	671-673	In the "Recommended Reference Materials for the Realization of Physicochemical Properties - Section: Absorbance and Wavelength" infrared absorption bands were suggested as reference for wavenumber calibration. More recent revised data have been published by IUPAC in a book entitled <u>Tables of Wavenumbers for the Calibration of Infrared Spectrometers</u> compiled by A.R.H. COLE (Pergamon Press, Oxford, 1977). Wavenumbers appearing in sections II/8, II/9, II/10 and II/11 of Recommendations on Reference Materials should be replaced by the values published by COLE.				
54, 1 (1982)	208, Abstract, line 2	<u>for</u> 'dexcy' <u>read</u> 'deoxy'				
	209, column 2, 3rd name	<u>for</u> 1,1,2,2-Tetrahydro <u>read</u> 1,1,2,2-Tetradehydro				
	210, column 1, 4th name	<u>for</u> 6,7,7,8-Tetrahydro <u>read</u> 6,7,7,8-Tetradehydro				
	214, column 2, structure for Cladinose	<table style="width: 100%; border: none;"> <tr> <td style="text-align: center; vertical-align: middle;"><u>for</u></td> <td style="text-align: center; vertical-align: middle;"> <math display="block">  \begin{array}{c}  \text{CHO} \\    \\  \text{CH}_2 \\    \\  \text{CH}_3\text{OC}-\text{CH}_3 \\    \\  \text{HOCH} \\    \\  \text{CH}_3  \end{array}  </math> </td> <td style="text-align: center; vertical-align: middle;"><u>read</u></td> <td style="text-align: center; vertical-align: middle;"> <math display="block">  \begin{array}{c}  \text{CHO} \\    \\  \text{CH}_2 \\    \\  \text{CH}_3\text{OC}-\text{CH}_3 \\    \\  \text{HOCH} \\    \\  \text{HOCH} \\    \\  \text{CH}_3  \end{array}  </math> </td> </tr> </table>	<u>for</u>	$  \begin{array}{c}  \text{CHO} \\    \\  \text{CH}_2 \\    \\  \text{CH}_3\text{OC}-\text{CH}_3 \\    \\  \text{HOCH} \\    \\  \text{CH}_3  \end{array}  $	<u>read</u>	$  \begin{array}{c}  \text{CHO} \\    \\  \text{CH}_2 \\    \\  \text{CH}_3\text{OC}-\text{CH}_3 \\    \\  \text{HOCH} \\    \\  \text{HOCH} \\    \\  \text{CH}_3  \end{array}  $
<u>for</u>	$  \begin{array}{c}  \text{CHO} \\    \\  \text{CH}_2 \\    \\  \text{CH}_3\text{OC}-\text{CH}_3 \\    \\  \text{HOCH} \\    \\  \text{CH}_3  \end{array}  $	<u>read</u>	$  \begin{array}{c}  \text{CHO} \\    \\  \text{CH}_2 \\    \\  \text{CH}_3\text{OC}-\text{CH}_3 \\    \\  \text{HOCH} \\    \\  \text{HOCH} \\    \\  \text{CH}_3  \end{array}  $			
54, 3 (1982)	681, Abstract, line 3	<u>for</u> TiO <sub>2</sub> 1892±30 <u>read</u> TiO <sub>2</sub> 1843±15 (in air) <u>for</u> ZnO <sub>2</sub> 2710±25 <u>read</u> ZnO <sub>2</sub> 2710±35				
54, 8 (1982)	1455, Table 1 (i) column -ΔH <sub>h</sub> <sup>o</sup> for Hg <sup>2+</sup>	<u>for</u> 1940 <u>read</u> 1845				
	(ii) footnote a to Table 1	<u>add</u> ΔH <sub>h</sub> <sup>o</sup> of Hg <sup>2+</sup> has been misprinted in Ref. 19				
	1457, Table 2 column 1, last item	<u>for</u> BP <sub>4</sub> <sup>-</sup> <u>read</u> BPh <sub>4</sub> <sup>-</sup>				
	1459, Table 3 column 1, 12th item	<u>for</u> Hg(l)/Hg <sup>2+</sup> <u>read</u> Hg(l)/Hg <sub>2</sub> <sup>2+</sup>				
54, 10 (1982)	1859, Eq.(14)	correct version is as follows:				

